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Spectroscopy Letters

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713597299>

Determination of the Stereochemistry of Substituted 4-(Sulfo- and Sulfonamidoalkyl) piperidine-2-carboxylic Acids with ¹H NMR, COSY, and Homonuclear NOE Experiments

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To cite this Article hadri, Ahmed El , Thomasson, François and Leclerc, Gérard(1995) 'Determination of the Stereochemistry of Substituted 4-(Sulfo- and Sulfonamidoalkyl) piperidine-2-carboxylic Acids with ¹H NMR, COSY, and Homonuclear NOE Experiments', Spectroscopy Letters, 28: 5, 795 — 803

To link to this Article: DOI: 10.1080/00387019508009920

URL: <http://dx.doi.org/10.1080/00387019508009920>

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**DETERMINATION OF THE STEREOCHEMISTRY OF
SUBSTITUTED 4-(SULFO- AND SULFONAMIDOALKYL)
PIPERIDINE-2-CARBOXYLIC ACIDS WITH ^1H NMR, COSY, AND
HOMONUCLEAR NOE EXPERIMENTS**

Key Words : NMDA, competitive antagonists, (sulfo- and sulfonamidoalkyl) piperidine-2-carboxylic acids, ethyl 4-(hydroxyalkyl)piperidine-2-carboxylates, ^1H NMR spectra, Homonuclear NOE.

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ABSTRACT

Basically the aim of this work is to define the accurate configuration of 4-substituted (sulfo- and sulfonamidoalkyl)piperidine-2-carboxylic acids which have been conceived as potential NMDA antagonists. ^1H NMR and 2D NMR (COSY)

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followed by qualitative Homonuclear NOE have led to the assignment of the (\pm) cis and (\pm) trans configuration of the whole family's compounds.

INTRODUCTION

Numerous NMDA receptor antagonists have been described these last years^{1,2}. Several of them were found active both against epilepsy³, ischemia⁴ and as neuroprotectors⁵. CGS-19755 [4-(phosphonomethyl)piperidine-2-carboxylic acid]⁶ and LY-257883⁷ were among the most potent and promising compounds.

We have focussed our attention on sulfonic analogs e.g. 4-(sulfo- and sulfonamido alkyl)piperidine-2-carboxylic acid derivatives as potential NMDA receptors antagonists^{8,9}(fig.1). The key compounds in the preparation of this family were the ethyl-4-(hydroxy alkyl)piperidine-2-carboxylates (scheme I).

These compounds were obtained by catalytic reduction of the corresponding ethyl 4-(hydroxyalkyl)pyridine-2-carboxylate in AcOH with PtO_2 as catalyst. Unexpectedly enough both (\pm) cis and (\pm) trans isomers were obtained for $n = 1$ and 2.

^1H NMR and COSY experiments permitted to attribute the stereochemistry of compounds **1a** and **1b**, however for **2a** and **2b** additional Homonuclear NOE were needed.

EXPERIMENTAL

The ^1H NMR spectra and NOE experiments were recorded at 20°C in 5 mm tube on a Bruker AC 200 spectrometer with a proton operating frequency of 200.13 Mhz and referenced to the CDCl_3 or D_2O signal.

Experiments in NOE difference spectroscopy were achieved on samples prealably filtered and degassed. The NOE is measured according to the method of Bell and Saurders¹⁰.

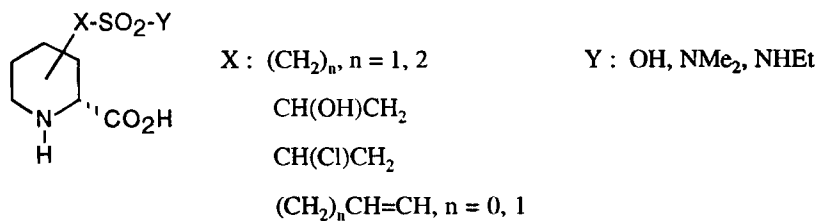
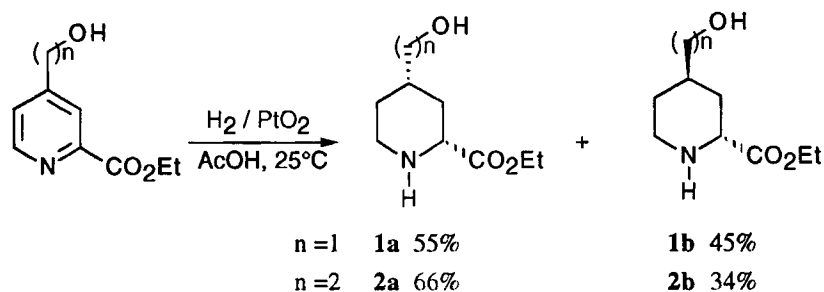


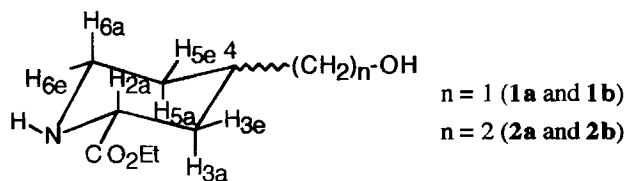
FIG. 1.



Scheme I.

TABLE 1.

Chemical Shifts ^1H δ (ppm) of Compounds **1a**, **1b**, **2a** and **2b**.



Compd	H _{2a}	H _{3a}	H _{3e}	H ₄	H _{5a}	H _{5e}	H _{6a}	H _{6e}
1a	3.02	0.95	1.49	1.96	0.95	1.49	2.61	3.20
1b	3.12	0.92	1.57	1.97	1.05	1.57	2.60	3.28
2a	3.29	1.07	2.04	1.65	1.07	1.65	2.61	3.13
2b	3.29	1.07	2.04	1.65	1.07	1.65	2.61	3.13

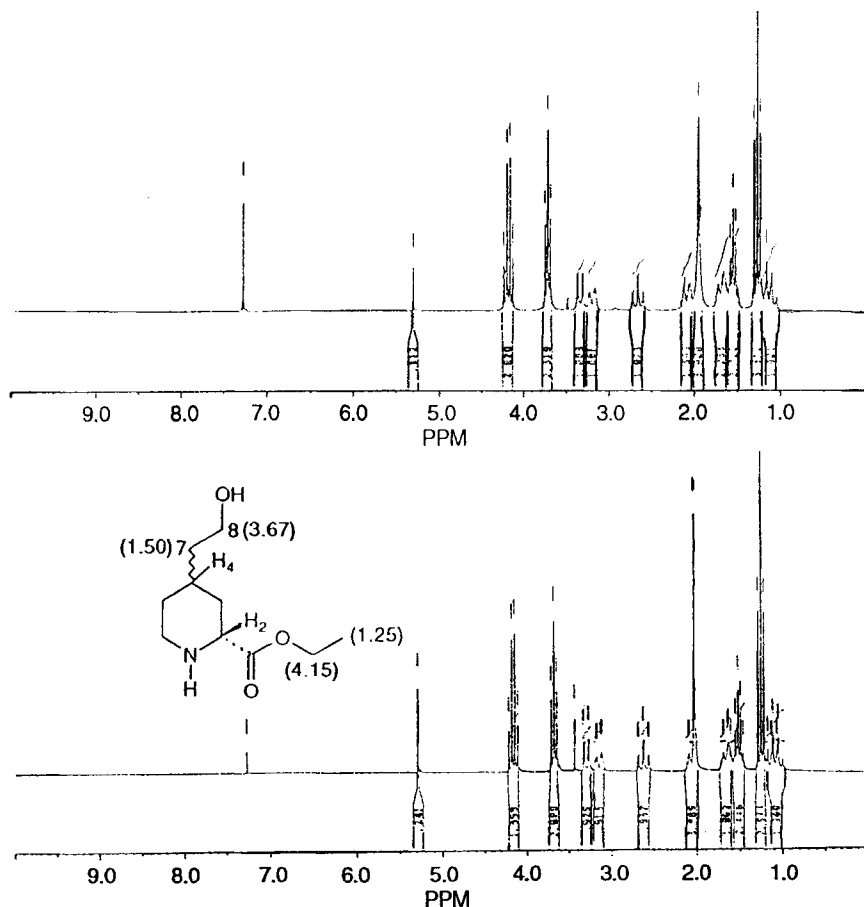


FIG. 2. ^1H NMR Spectra of Compounds 2a and 2b in CDCl_3

RESULTS AND DISCUSSION

The assignment of (\pm) *cis* and (\pm) *trans* configurations of compounds 1a, 1b, 2a and 2b was based on ^1H NMR spectra assuming a chair conformation for these derivatives. Selective irradiations were helpful to achieve the accurate configuration of all protons (table 1, fig.2).

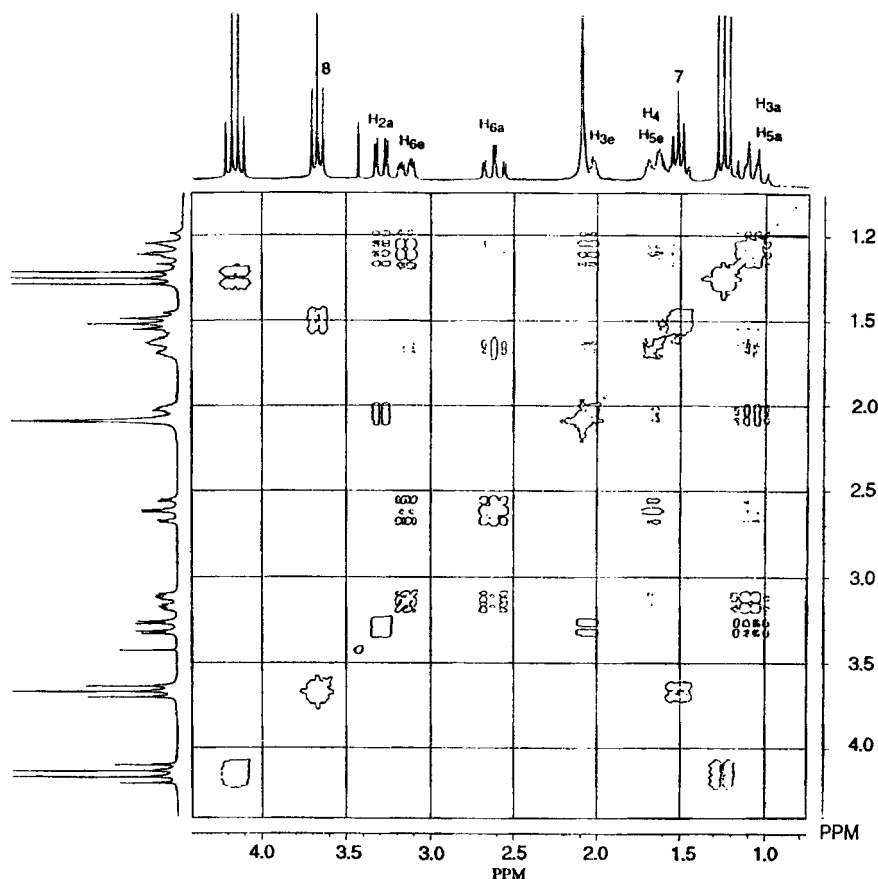


FIG. 3. COSY Spectra of Compound **2b**

2D ^1H NMR were in agreement with the initial interpretations. The COSY spectrum of compounds **2b** (fig.3) is given as an example.

- Coupling constants measured on the H_2 proton ($J = 12.0, 3.0$ Hz) of compounds **1a**, **1b**, **2a** and **2b** indicate an axial position.

- coupling constants measured at 1.96 and 1.97 ppm for **1a** ($J = 12.02, 3.0$ Hz) and **1b** ($J = 3.0, 2.2$ Hz) indicate respectively an axial and an equatorial position

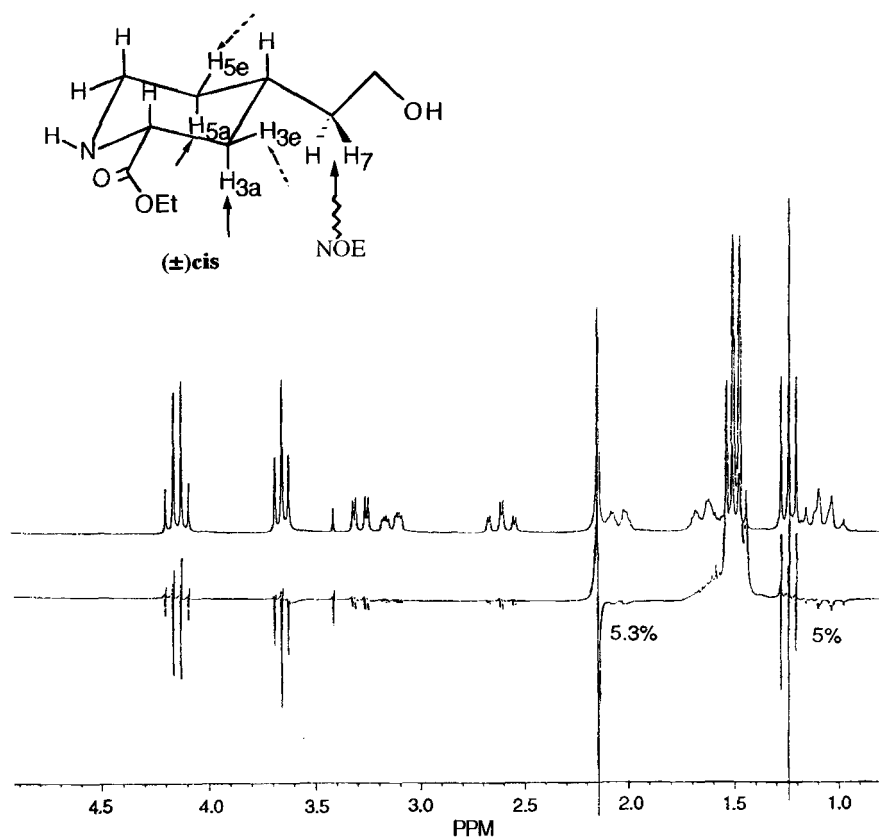
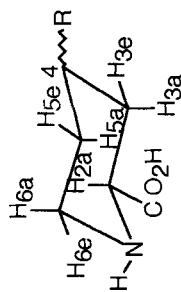


FIG. 4.

for H₄ and consequently a (±) cis configuration for **1a** and a (±) trans configuration for **1b**.

- the signal at 1.65 ppm for **2a** or **2b** corresponds to protons H₄ and H_{5e}. The coupling constant associated to this signal ($J = 12\text{--}13\text{ Hz}$) did not permit to attribute the configuration of H₄, because the geminal coupling between H_{5a} and H_{5e} and the vicinal H_{4a}, H_{5a} coupling are identicals. This assignment was achieved by Homonuclear NOE. Selective irradiations of H₇ protons of **2a** and **2b** indicate a

TABLE 2.

Chemical Shifts δ ^1H (ppm) of the Disubstituted-2,4-Piperidine Derivatives.

Compd	R	Stereochem.	H _{2a}	H _{3a}	H _{3e}	H ₄	H _{5a}	H _{5e}	H _{6a}	H _{6e}
3a	CH ₂ SO ₃ H	(±)cis	3.63	1.32	2.36	2.02	1.32	2.02	2.87	3.33
3b	(CH ₂) ₂ SO ₃ H	(±)cis	3.37	1.15	2.11	1.72	1.15	1.72	2.79	3.25
3c	(CH ₂) ₂ SO ₂ NMe ₂	(±)cis	3.50	1.18	2.14	1.80	1.18	1.80	2.80	3.27
3d	CH(OH)CH ₂ SO ₂ NHEt	(±)cis	3.07	0.88	1.79	1.44	0.88	1.44	2.32	2.88
3e	CH(Cl)CH ₂ SO ₂ NHEt	(±)cis	3.32	1.31	2.12	1.71	1.31	1.71	2.68	2.91
4a	CH=CH-SO ₂ NMe ₂ *	(±)cis	3.75	1.42	2.26	1.89	1.42	1.89	2.87	3.34
4b	CH ₂ CH=CH-SO ₂ NMe ₂ *	(±)cis	3.52	1.14	2.32	2.01	1.14	2.01	2.79	3.22
4c	CH ₂ CH=CH-SO ₂ NMe ₂ *	(±)trans	3.56	1.16	2.33	2.03	1.16	2.03	2.81	3.24

* : Z configuration

mutual relationship between H_{3a} , H_{5a} and H_{3e} for **2b** (NOE effect of 5% for H_{3a} , H_{5a} and 5,3% for H_{3e} , for H_{5e} the NOE effect could not be observed because of its vicinity with H_7). These data are only in agreement with a (\pm) cis stereochemistry (fig 4).

On the contrary irradiation of H_7 for **2a** has no effect on H_{3a} , H_{5a} and H_{5e} and only a weak one on H_{2a} and H_{6a} . Consequently the stereochemistry of **2a** is (\pm)trans.

The above results were used to assign unambiguously the accurate stereochemistry of a series of compounds derived from **1a**, **1b**, **2a** and **2b**. The data are gathered in table 2.

CONCLUSION

Basically the aim of this study is to define the exact configuration of disubstituted-2,4-piperidines. A previsual survey has been undertaken from the 1H and 2D spectra analysis. The qualitative study with Homonuclear NOE effect leads to the assignment of the accurate configuration and reveal a good coherence with the previsual study.

We have also observed that in the all series of 2,4-disubstituted piperidines, the shielding of piperidine protons varies in the decreasing order of $H_{2a} > H_{6e} > H_{6a} > H_{3e} > H_4$, $H_{5e} > H_{3a}$, H_{5a} . Consequently this observation can be used as an easy method to assign the configuration of these compounds.

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Date Received: February 23, 1995

Date Accepted: March 22, 1995